

Application No. 09/964,905  
Examiner-GAU Bell - 1746

Prepared by Lois Stone  
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## JACKET

a. Serial No.	f. Foreign Priority	k. Print Claim(s)	p. PTO-1449
b. Applicant(s)	g. Disclaimer	l. Print Fig.	q. PTOL-85b
c. Continuing Data	h. Microfiche Appendix	m. Searched Column	r. Abstract
d. PCT	i. Title	n. PTO-270/328	s. Sheets/Figs
e. Domestic Priority	j. Claims Allowed	o. PTO-892	t. Other

## SPECIFICATION

- a. Page Missing
- b. Text Continuity
- c. Holes through Data
- d. Other Missing Text
- e. Illegible Text
- f. Duplicate Text
- g. Brief Description
- h. Sequence Listing
- i. Appendix
- j. Amendments
- k. Other

## CLAIMS

- a. Claim(s) Missing
- b. Improper Dependency
- c. Duplicate Numbers**
- d. Incorrect Numbering
- e. Index Disagrees
- f. Punctuation
- g. Amendments
- h. Bracketing
- i. Missing Text
- j. Duplicate Text
- k. Other

## MESSAGE

There are two original claims numbered 79,  
in the claims dated 9/27/2001.  
Please advise.

Thank you,

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## RESPONSE

initials

73. An electrode as recited in claim 59, wherein said second reactive material comprises lead-based compounds for use within a lead-acid liquid electrolyte battery.

74. An electrode as recited in claim 59, wherein said second reactive material comprises a mixture of lead-oxide, glass fibers, and sodium per borate.

75. An electrode as recited in claim 59, wherein said second reactive material comprises a mixture of sulfates, hydroxides, free lead, carbonates, and a binding agent.

76. A method of increasing chemical reaction efficiency for an electrode assembly configured for use within a liquid electrolyte battery, comprising:  
forming a chamber within a first active material; and  
inserting a highly reactive second material within the chamber, wherein the highly reactive second material is capable of supporting charge generation within the liquid electrolyte battery.

77. A method as recited in claim 76, wherein the highly reactive second material comprises a non-structural material which provides a higher per-unit area reaction efficiency than that of the first active material.

78. A method as recited in claim 76, wherein the highly reactive second material comprises a reactive material configured in a particulated form which increases reactive surface area.

79. A method as recited in claim 76, wherein the highly reactive second material comprises lead-based compounds for use within a lead-acid liquid electrolyte battery.

79. A method as recited in claim 76, wherein said reactive material is created from mixing a composition comprising lead-oxide, glass fibers, and

sodium per borate.

80. A method as recited in claim 76, wherein the reactive material is created from mixing a composition comprising a mixture of sulfates, hydroxides,  
5 free lead, carbonates, and a binding agent.